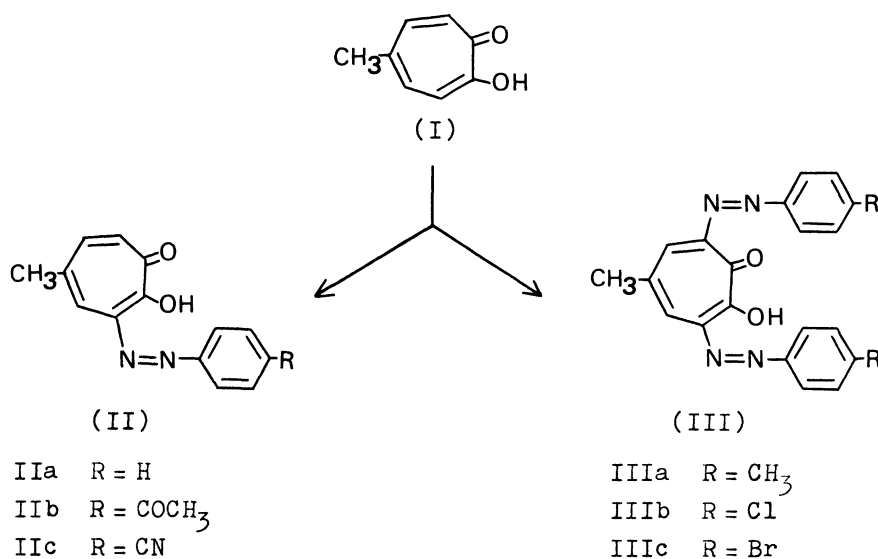


DIAZO-COUPLING REACTION OF 5-METHYLTROPOLONE.
THE FORMATION OF 3,7-DIARYLAZOTROPOLONES

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5-Methyltropolone reacted with aryldiazonium salts to give either 3-arylaazo-5-methyltropolones or 3,7-diarylaazo-5-methyltropolones according to the electrophilicity of the attacking species.

It is well known that tropolones having no substituent at the C-5 position give nicely crystalline 5-arylazotropolones by the diazo-coupling reaction.¹⁾ In the case of 5-substituted tropolones, there are the following two examples with regard to the reactions with aryldiazonium salts. Firstly, a tropolone substituted at the C-5 position with chlorine, isopropyl, methoxyl, or phenyl group gave the corresponding 3-arylazotropolone when treated with an aryldiazonium salt.²⁾ Secondly, tropolones having hydroxymethyl,³⁾ morpholinomethyl,⁴⁾ troyl,⁵⁾ bromine,²⁾ or iodine²⁾ at the C-5 position underwent replacement of the substituent with arylazo group. Now, we wish to report the third example, the formation of 3,7-diarylazotropolones (III), of diazo-coupling reaction of troponoid compounds, as shown in Scheme 1.



Scheme 1

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When 5-methyltropolone (I) was treated with p-acetylbenzenediazonium chloride having a strongly electron-withdrawing group at the para position, at low temperature ($-5 \sim -10^\circ\text{C}$), 3-(p-acetylphenylazo)-5-methyltropolone (IIb) was obtained: yield 68%, mp 163°C ; IR (KBr) $1700(\text{C}=\text{O})$, $1595(\text{C}=\text{O})$, 1555 cm^{-1} ; NMR (CF_3COOH) δ 2.46(s, 3H, COCH_3), 2.66(s, 3H, CH_3), 7.29(s, 2H, H-4,6), 7.53(s, 1H, H-7), 7.93(d, 2H, H \times 2, $J=9.2\text{Hz}$), 8.13(d, 2H, H \times 2, $J=9.2\text{Hz}$); found C 67.69, H 5.01, N 9.81%, calcd for $\text{C}_{16}\text{H}_{14}\text{N}_2\text{O}_3$ C 68.07, H 5.00, N 9.92%; quinoxaline derivative, mp 272°C . The reaction with p-cyanobenzenediazonium salt gave also a 3-arylazotropolone derivative (IIc): yield 71%, mp 182°C .

On the other hand, when p-tolyldiazonium salt was used under the same condition, 3,7-diarylaazo-5-methyltropolone (IIIa) was obtained: yield 15%, mp 183°C ; IR (KBr) $1602(\text{C}=\text{O})$, 1545 cm^{-1} ; NMR (CF_3COOH) δ 2.67(s, 6H, $\text{CH}_3\times 2$), 2.76(s, 3H, CH_3), 7.55(d, 4H, H \times 4, $J=7.8\text{Hz}$), 7.97(d, 4H, H \times 4, $J=7.8\text{Hz}$), 8.56(s, 2H, H-4,6); found C 70.57, H 5.48, N 14.53%, calcd for $\text{C}_{22}\text{H}_{20}\text{N}_4\text{O}_2$ C 70.95, H 5.41, N 15.05%; quinoxaline derivative, mp 183°C . The reaction of an aryldiazonium salt having a weakly electron-withdrawing (Cl or Br) substituent gave also 3,7-diarylaazo-5-methyltropolone (III): IIIb, yield 63%, mp 205°C ; IIIc, yield 72%, mp 207°C .

In addition, the reaction with benzenediazonium chloride yielded a tarry material which gave a quinoxaline derivative (mp 172°C , 38%) of the form (II) with o-phenylenediamine. The structure of all the products was confirmed by their spectral data and elemental analyses.

To summarize, the reaction seems to depend on both the substituent at the C-5 position and the electrophilicity of the attacking species. The details will be reported in the near future.

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